

Correlated electrons and complicated integrals

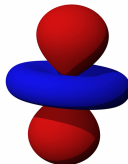
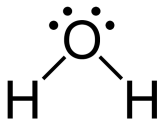
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Chemistry is fundamentally about the behaviour of electrons
e.g. reaction, bonding, structure, catalysis



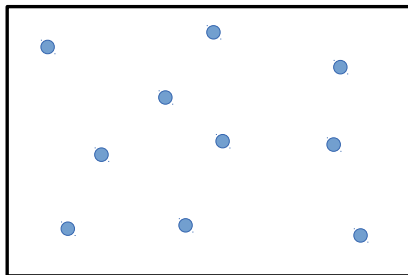
$$H\Psi = E\Psi$$

- Atoms and molecules - lots of electrons.
- Negatively charged electrons repel one another.
- Molecular electronic structure - **a many body problem.**

Many-body problems

Definition

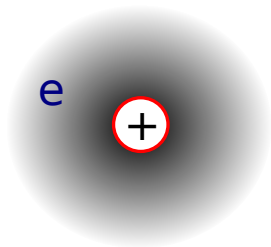
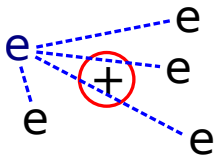
“the study of the effects of interactions between bodies on the behaviour of a many-body system”



Many non-interacting particles \rightarrow many 1-body problems.
BUT interactions are often important to describe properties.

Hartree-Fock theory

- Each electron sees an average field of all other electrons.
- Approximate solution of the Schrödinger equation.
- Account for $\sim 99\%$ of E_{exact} .

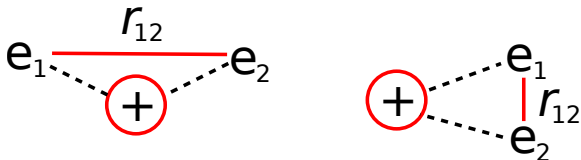


BUT some electron-electron interaction is neglected.

Electron correlation

$$E_{\text{exact}} = E_{\text{HF}} + E_{\text{corr}}$$

- Electronic repulsive interaction $\sim \frac{1}{r_{12}}$.
- Electronic motion is “correlated”.



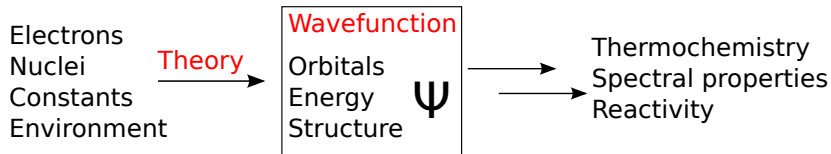
- Electron correlation is important to describe chemistry!
 - Bond dissociation
 - Dispersion interactions

e.g. E_{corr} for H_2 is $\sim -100 \text{ kJ mol}^{-1}$.

Approximate wavefunctions

Wavefunction

A mathematical object which fully describes the the state of a quantum mechanical system.



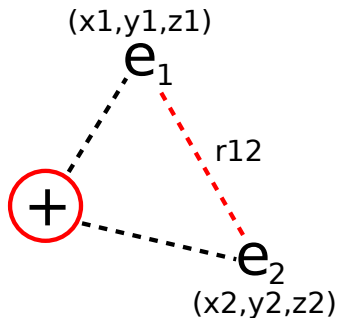
- We want to find Ψ , but this is hard!

$$\hat{H}\Psi(\mathbf{x}_1, \mathbf{x}_2, \dots, \mathbf{x}_N) = E\Psi(\mathbf{x}_1, \mathbf{x}_2, \dots, \mathbf{x}_N)$$

- Build approximate Ψ from a **basis set** of simpler functions.

Explicitly-correlated methods

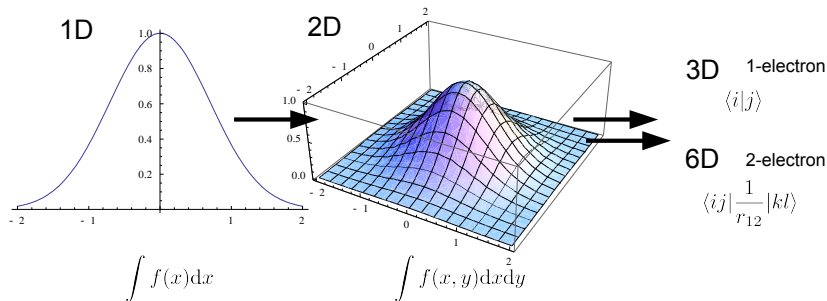
- Build upon Hartree-Fock theory.
- Efficiently describe electron correlation.
- Introduce functions with explicit dependence on r_{12} .
- More accurate E_{corr} with a smaller basis set.



BUT: Complicated “many-electron” integrals arise.

Tricky integrals

- All methods require 1- and 2- electron integrals.



- There are efficient methods for computing these.
- Many-electron (> 2) integrals in explicitly correlated methods are numerous and complicated to evaluate.

Approximations are required for practical calculations!

Resolutions of the identity (RIs)

- Widely used (R12/F12 methods).
- Introduces an additional set of functions (RI basis).
- Many electron integrals are broken down into more manageable 2-electron integrals, e.g.

$$\langle ij m | r_{12}^{-1} F_{23} | m k l \rangle \approx \sum_p \langle ij | r_{12}^{-1} | m p \rangle \langle m p | F_{23} | k l \rangle$$

1 × 3-electron integral → sum over 2 × 2-electron integral

Problem: A good approximation may require a large RI basis with unwieldy high-angular momentum functions.

Kutzelnigg, W. *Theor. Chim. Acta.* 68, 445-469 (1985).

Kutzelnigg, W. & Klopper, W. *J. Chem. Phys.* 94, 1985 (1991).

Density-fitting (DF)

- Established technique in other contexts.
- Introduces an additional set of functions (DF basis).
- Complicated many-electron integrals are approximated as sums over simpler many-electron integrals, e.g.

$$(im|r_{12}^{-1}|jk|F_{23}|ml) \approx \sum_{A,B,C} D_A^{im} D_B^{jk} D_C^{ml} (A|r_{12}^{-1}|B|F_{23}|C)$$

3-electron, 6-index integral \rightarrow sum over 3-electron, 3-index integrals

Advantage: A good approximation with a smaller DF basis.

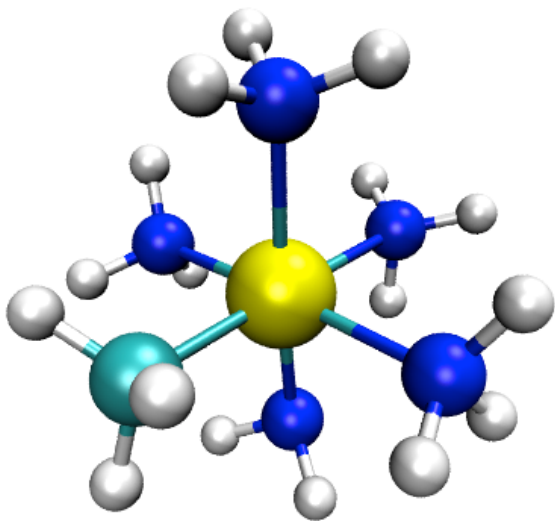
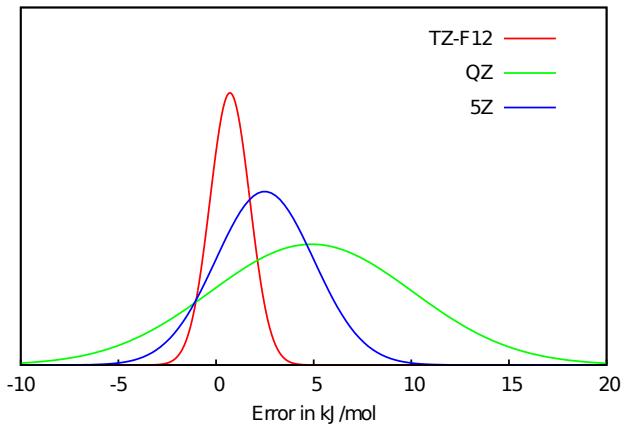


Image: Vitamin B12 model, $[\text{Co(III)}(\text{NH}_3)_4\text{NH}_2\text{CH}_3]^+$, Patrick von Glehn.

CCSD vs CCSD-F12 reaction energy errors for a set of 50 test reactions.



Plot: Dr. David Tew, University of Bristol

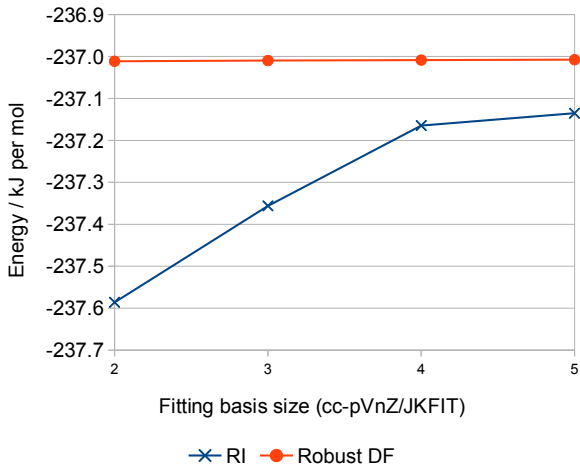
Aim

Perform explicitly-correlated (MP2-F12) calculations using density-fitting in the places of resolutions of the identity.

- Derivation of relevant integral forms.
- Development of computer software to generate integrals.
- Interface with existing MP2-F12 software in **MOLPRO**.
- Run test calculations to compare approximation methods.

MOLPRO, a package of ab initio programs, H.-J. Werner, P. J. Knowles, G. Knizia, F. R. Manby, M. Schütz, and others , see <http://www.molpro.net>.

V-matrix contribution to MP2-F12 energy of H2O



MP2-F12/3*A(DX, FIX) energies calculated in MOLPRO. The RI was performed with an ABS.

Summary

- DF may be competitive with RIs for approximating many-electron integrals in MP2-F12 theory.
- Promising initial results:
 - Accurate integrals.
 - Fast convergence with basis set size.
- Proof-of-concept implementation in MOLPRO.

Further work

- A completely RI-free MP2-F12 calculation.
- Improve performance of software.

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